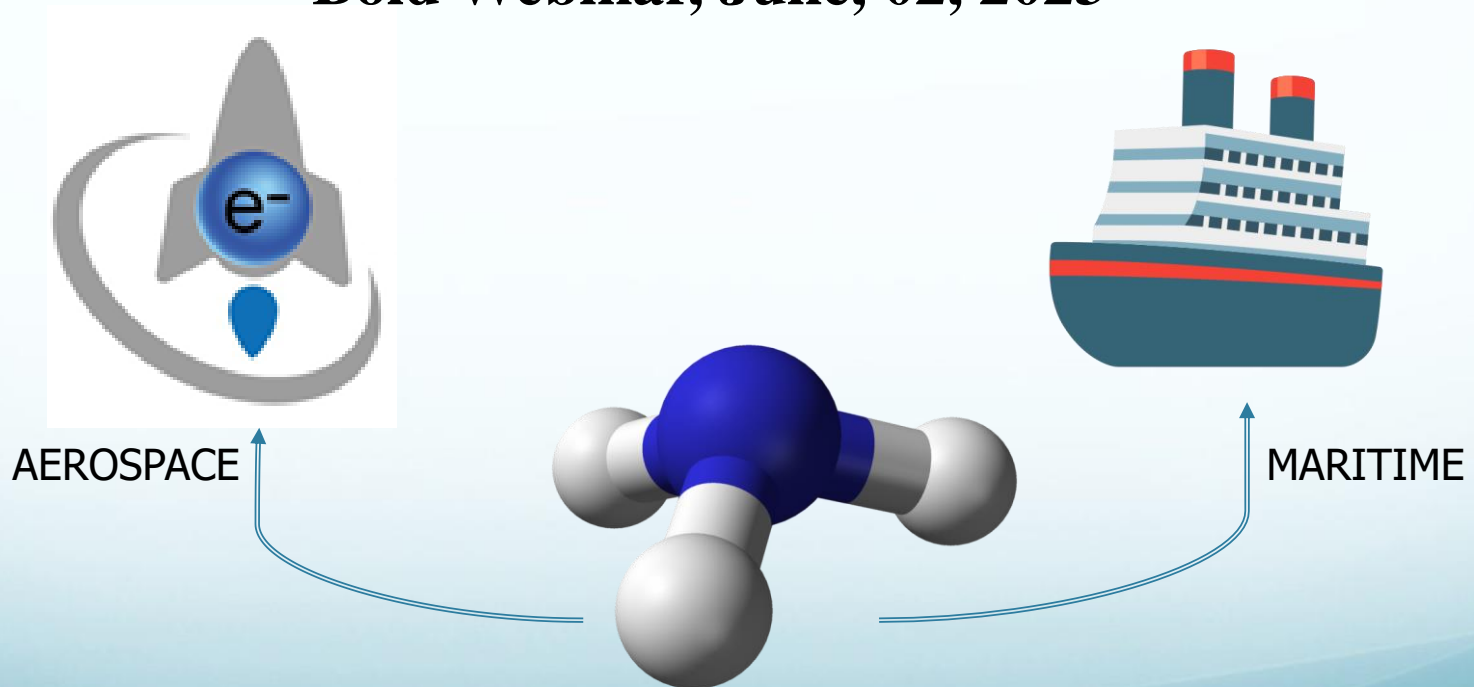


NH₃ cracking and pure H₂ extraction

Bold Webinar, June, 02, 2023



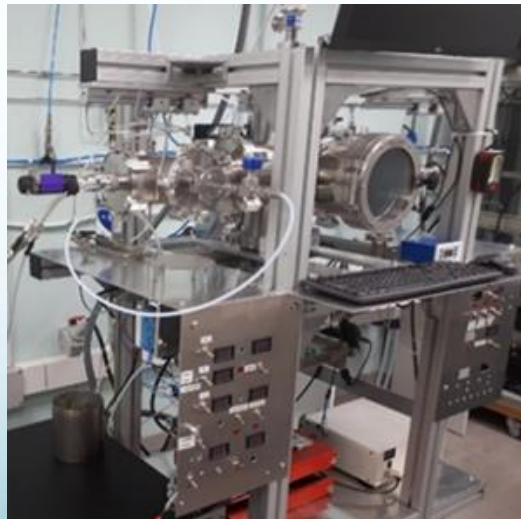
- 1.- ATD brief presentation.**
- 2.- NH₃ as H₂ Molecular Packer**
- 3.- NH₃ Cracking. Technology and features.**
- 4.- NH₃ cracker specs.**

1.- ATD presentation. Expertise and facilities.

ADVANCED THERMAL DEVICES (ATD)

Technological Spanish SME company initially focused on refractory materials and heat treatments.

During the last seven years there has specialized in electro ceramics, especially for the emission of electrons applied to electric propulsion for the aerospace industry. So, ATD is the NEMESIS project leader (European project ID:870506, <https://cordis.europa.eu/project/id/870506>).



1.- ATD presentation. Recent Projects.



Financiado por la
 Unión Europea
 NextGenerationEU



Plan de Recuperación,
 Transformación y Resiliencia



Oct,2022 – Dec,2024. COMPLETE AMMONIA CHAIN.
 NH3 Synthesis (No Haber-Bosch. Catalysts reactor).
 NH3 Storage and logistics
 NH3 cracking on board. Pure H2 extraction:

- H2 PEM FC
- NH3 H-SOFC

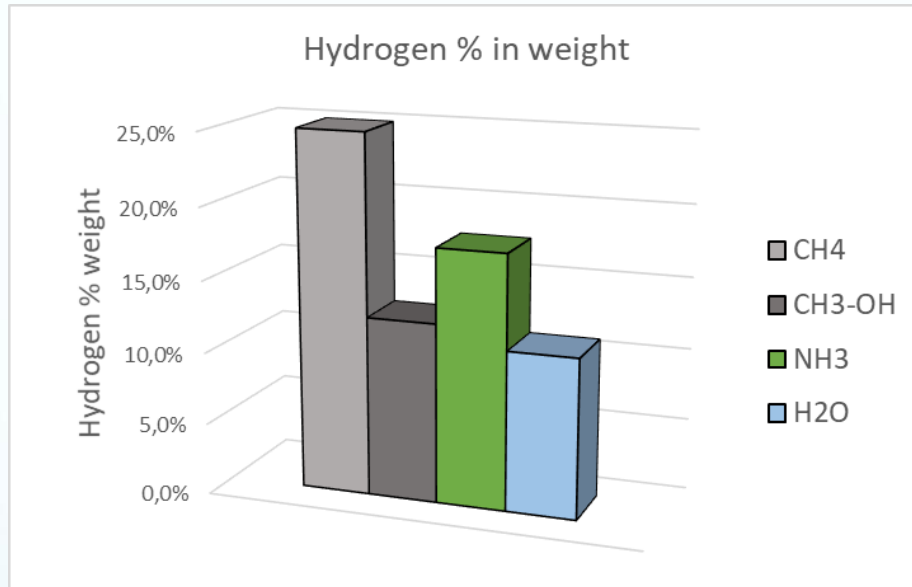


Oct,2022 – Dec,2024
 Zero emission and high autonomy UAVs
 New VTOL concept: plane+dron
 H2 PEM FC
 NH3 H-SOFC



2.- NH3 as H2 molecular packer

NH3 is the second best molecular H2 packer (behind CH4) and the FIRST Carbon-free H2 packer.



	MAXIMUM H2 MOLECULAR PACKAGING			
	Methane	Methanol	Ammonia	Water
	CH4	CH3-OH	NH3	H2O
Mol weigh	16	32	17	18
H weight/mol	4	4	3	2
%H in weight	25,0%	12,5%	17,6%	11,1%

1 Kg NH3 (liquid) = 1.6 L at 10bar@25°C or 1 bar@-33°C.

1 Kg NH3 contains 176 gr H2 or 5,88 KWh energy in terms of H2

1 Kg NH3 (liq) (1.6L) has around 850 compression ratio vs. gas, that means. That means an energy density higher than the same volume of 700bar H2, that is:

- 1971 L of H2 1bar
- 2.8 L of H2 700 bar

1Kg NH3 (liq) (1.6 L) = H2 (1.6 L) 1235 bar, that means, the same volume (1.6 L) implies an H2 compression about 1235 bar!!!!

2.- NH3 Cracking. Basics.

Basic reaction: $2\text{NH}_3 + 92\text{KJ/mol} \leftrightarrow 3\text{H}_2 + \text{N}_2$

NH3 cracking can be implemented by:

- **Temperature as most relevant parameter** (with support from basic catalysts). Temperatures 500°C-700°C
- **Advanced catalysts.** Lower temperatures and higher dissociation ratio (reaction tends to the right, dissociation).

Dissociation implies a minimum energy 92KJ/mol that represent around 13% in terms of H2 content. In other words, maximum energy efficient 87% without losses and 100% cracking reaction.

TARGETS:

- **30% maximum losses in terms of H2 content, that means 70% output H2.**
- **Lower temperature → lower power consumption**
- **To favor cracking: capture H2 to provoke more NH3 dissociation**

SOLUTIONS:

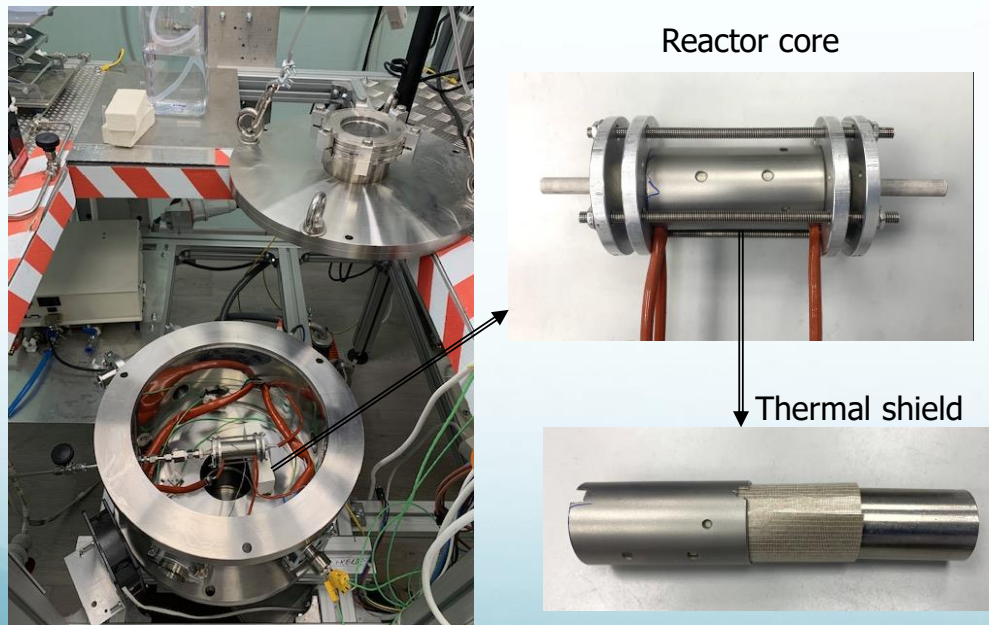
- **New catalysts materials and concepts. Electrodynamics vs. passive (temperature) catalysts (Patented). LOW TEMPERATURE AND HIGH DISSOCIATION RATIO.**
- **Thermal Energy Confinement. (High reflectance patents). LOW THERMAL LOSSES. More than 70% of thermal losses saved.**

3.- NH3 Cracking. Solutions.

SOLUTIONS: New catalysts concepts and very low thermal energy losses.

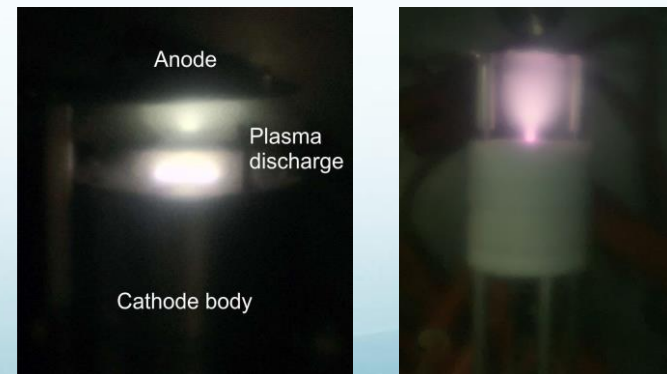
Catalyst based on N₂ and H₂ adsorption, ionization and neutralization. Materials with controlled bandgap allow **charge separation** (catalysts bed based on high temp semiconductor), complementary adsorption-ionization materials (N₂→2N⁺, H₂→ 2H⁻) in addition with high electrical conductivity allows an easy charge balance and ions neutralization. **Pulsed bias control (Patented).** **Low temperature operation (200°C-400°C).**

Low thermal losses: minimum thermal conduction losses and **thermal energy radiated confined** by high reflectance shields and **very low emittance by the hot nucleus (Patented).**



Background

First plasma neutralizer for space propulsion NH₃ fueled, presented in the International Electric Propulsion Conference (IEPC), jun-2022, MIT, Ma. (USA)

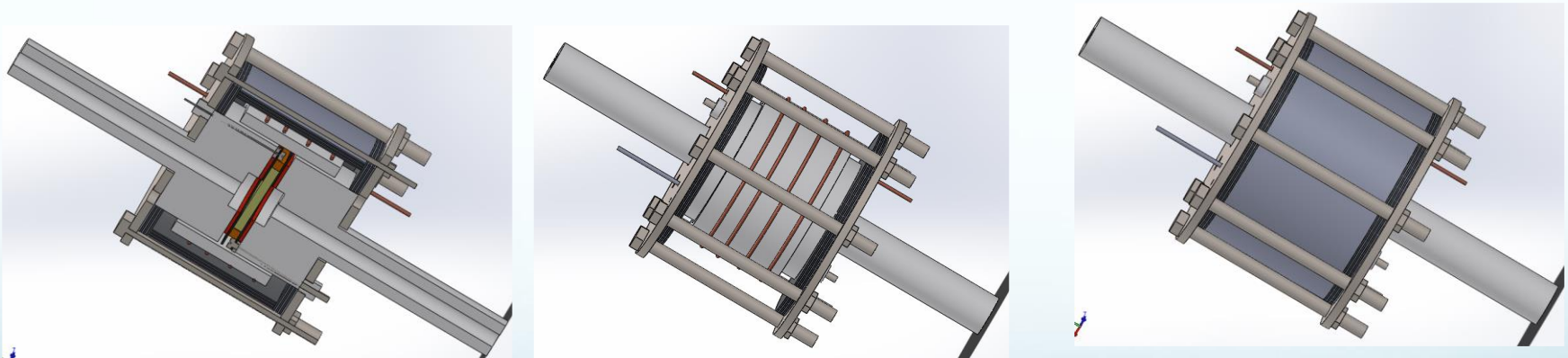


3.- NH3 Cracking. Other features.

PURE H2 EXTRACTION AND SOLID STATE PUMPING SYSTEM

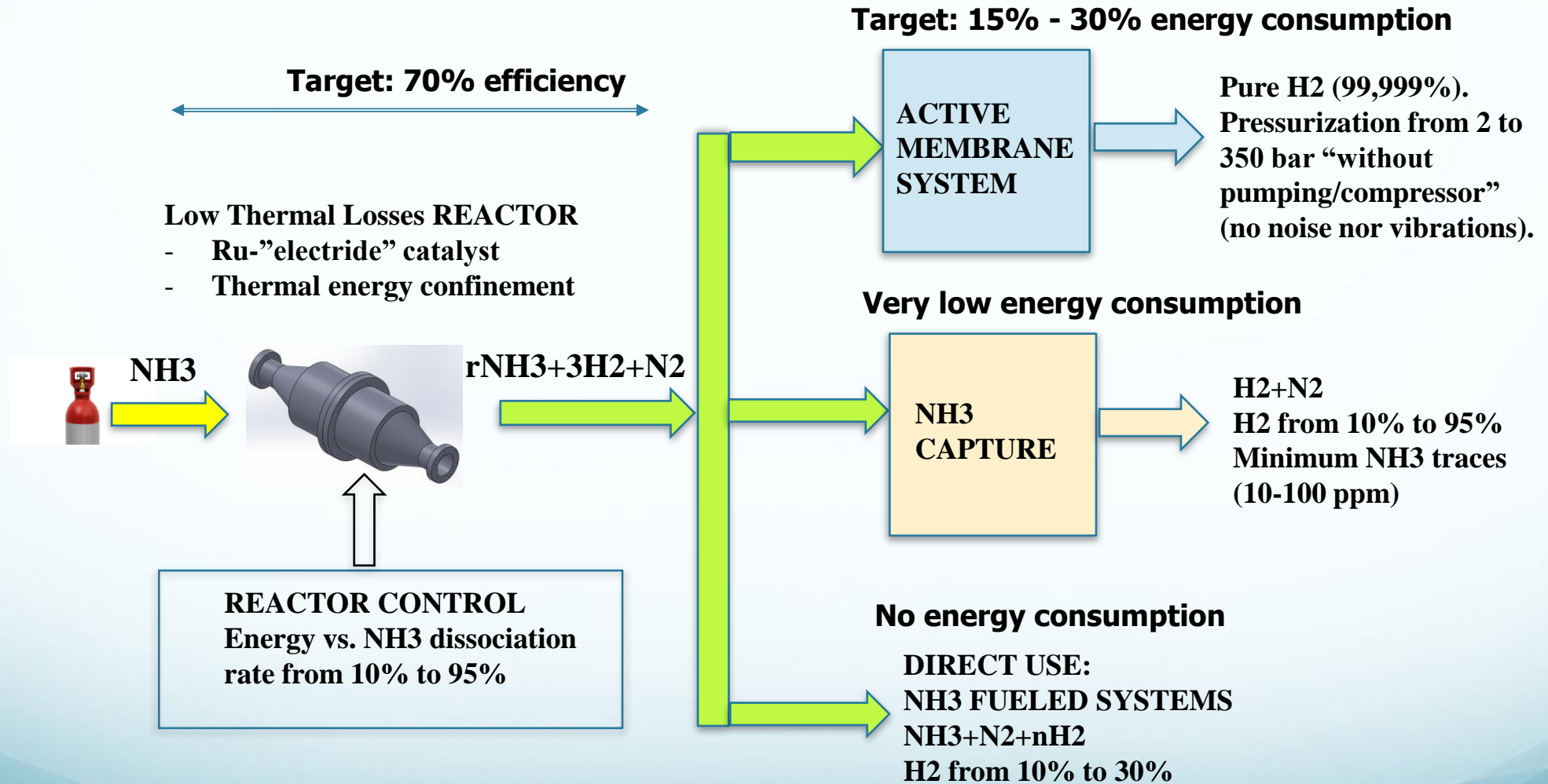
H2 EXTRACTION through an **ACTIVE PROTON MEMBRANE (APM)**, a ceramic membrane developed by ATD with very high proton conductivity that provides:

- Extracting H2 provokes the reaction shifts to the right, increasing dissociation to replace the “H2 lack”.
- H2 purification at the higher levels: only protons (H+) can pass through the membrane. 99,999% H2 purity.,
- Extra feature: H2 pumping in the output without mechanical pumping/compressors. Up to 350 bar H2 pressure in the output.



NH3 dissociation at different grades (10%-99.999%), with or without H2 purification and pressurization: OPEN ARCHITECTURE.

3.- NH3 Dissociation. OPEN ARCHITECTURE



4.- NH3 Cracker specs.

	Energy (KWh) per Kg NH3 cracked		KWh output / KWh NH3 in input		
	ATD TARGET SPECS		15% cracking	95% cracking	99,999% purity
	KWh/Kg NH3	%			
Mínimum Reaction Energy (Thermodynamics)	0,764	13,0%			
Basic Cracking (State of the art)	1,2	20,4%			
Thermal Losses	0,2	3,4%			
Total Energy (without H2 extraction and purif)	1,4	23,8%			
Total Energy per Kg.NH3 in input (H2 content)	5,88	100,0%			
Energy available after cracking (KWh/Kg NH3)	4,48	76,2%			
Process efficiency (best case)	4,48	76,2%	88,6%	76,2%	51,2%
Real process efficiency expected	4,41	75%	88,8%	75,0%	50,0%
Real process efficiency worst case	4,116	70%	89,5%	70,0%	45,0%

PLANNED RELEASES	Year-Quarter	Comments
15% Cracking. < 10 KW	2024-2T	
15% Cracking > 100 KW	2024-4T	End HIDRAM Project
95% Cracking < 10 KW	2024-3T	
95% Cracking > 100 KW	2025-2T	
99,999% purity < 10 KW	2024-4T	End HIDRAM Project
99,999% purity > 100 KW	2025-4T	

Thanks a lot for your attention.